

PROCESSING OF CONVERTER SLAG IN THE PROCESS OF OBTAINING Fe-Ni IN NEWCO FERRONIKELI COMPLEX L.L.C. OF DRENAS

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ABSTRACT

The Investigation introduce processing of converters slag into the smelting process in electro- reduction furnaces with arc during the production of Fe-Ni in NewCo Ferronikeli Complex L.L.C. in Drenas. With industrial tests is proved, that this type of material can be smelted inside the electro- reduction furnaces without difficultly, indicating in this case one increase of recovery of nickel and a production of electric furnace slag with lower smelting point and better cast ability, improving smelting process of production Fe – Ni as well. Processing of converter slag into the smelting process, has effected in considerable decreasing of Limestone into the calcinations process in rotary kilns. As result of decreasing of using Limestone, in rotary kilns, consumption of black oil has been decreased and has been achieved production of calcine with temperature above 750 C. Except improvement to the metallurgical process of obtaining Fe-Ni, has been decreased the pollution of environment as well, producing less CO₂ in rotary kilns and electro-reduction furnaces.

Key words: electric arc furnace, converter slag, electric furnace slag, crude Fe-Ni, recycling, dissociation.

PËRMBLEDHJE

Punimi prezanton riqarkullimin e zgjyrës së konvertorëve në procesin e shkrirjes, në furra elektro – reduktuese me hark elektrik për prodhimin e Fe-Ni në shkretoren e ferronikelit të ri në Drenas. Provat industriale kanë dëshmuar se ky lloj materiali mundet të shkrihet pa vështirësi në furrat elektro-reduktuese duke ndikuar në rritjen e shkallës së shfrytëzimit të nikelit si dhe në prodhimin e zgjyrave të furrave elektrike me pikë shkrirje ma të ulët dhe me rrjedhshmëri ma të mirë, duke e lehtësuar procesin e shkrirjes së prodhimit të Fe-Ni. Rikthimi i zgjyrës së konvertorëve në procesin e shkrirjes, ka ndikuar edhe

në zvogëlimin e konsiderueshëm të përdorimit të sasisë së gurit gëlqeror në procesin e fërgimit në furra rrotulluese. Si rezultat i zvogëlimit të përdorimit të gurit gëlqeror, në furra rrotulluese, është zvogëluar konsumi i mazutit dhe është arritur të prodhohet materiali i fërguar me temperaturë mbi 750 °C. Përveç përmirësimit të procesit metalurgjik të përfitimit të Fe-Ni, është zvogëluar edhe shkalla e ndotjes së mjedisit, duke prodhuar më pak CO₂ në furra rrotulluese dhe furra elektro-reduktuese.

Fjalët kyçe: furrë me hark elektrik, zgjyrë e konvertorit, zgjyrë e furrës elektrike, Fe-Ni i papërpunuar, riqarkullim, disocim.

1. INTRODUCTION

The aim of this investigation was recycling of converter slag into the smelting process of obtaining Fe-Ni, in order to improve the metallurgical process of obtaining Fe-Ni and consuming this material which contain considerable amount of Ni and since 2007 was considering as sterile and valueless material and as such it has been unusable. Smelting test of converter slag has been carried out at the Smelting Complex in NewCo Ferronikeli Complex L.L.C.- Drenas.

2. CONVERTER SLAG AND ITS PRODUCTION

In order to eliminate S, Si, Cr, P,C and other impurities, as well dressing of Ni content, the crude ferronickel goes for refining in vertical converter by technical O₂. In the converter the impurities are eliminated on the basis of selective oxidation of iron by oxygen. Limestone the size 16 – 32 mm is used for desulphurisation and obtaining of slag.

Slag of converter after cooling it is subject to magnetic separation and after the electromagnetic separation, the nickel concentrate is returned back to the converter for smelting, unified with liquid metal. Remaining slag is returned back in the smelting process

mixing with ores. Productions parameters of Fe-Ni for operation period January – June 2009 are presented in tables 1, 2 and 3.

| Month | Charges | FeNi(tn) | Ni(tn) | Ni/charge | Nr of charges converter 1 | Nr of charges converter 2 | Recovery | Efectiv time(min)/charge | Total time(min)/charge |
|----------|---------|----------|---------|-----------|---------------------------|---------------------------|----------|--------------------------|------------------------|
| January | 106 | 1336.4 | 339.9 | 3.21 | 53 | 53 | 96.84 | 39.57 | 89.95 |
| February | 113 | 1457.2 | 354.9 | 3.14 | 64 | 49 | 96.2 | 40.22 | 92.61 |
| March | 120 | 1690.51 | 435.4 | 3.63 | 75 | 45 | 98 | 40.05 | 86.88 |
| April | 118 | 1722.9 | 452.8 | 3.84 | 50 | 68 | 98 | 46.19 | 103.92 |
| May | 121 | 1875.0 | 454.5 | 3.76 | 74 | 47 | 97.43 | 38.66 | 80.6 |
| June | 140 | 2201.8 | 554.5 | 3.96 | 78 | 62 | 101.8 | 40.9 | 106.9 |
| Total | 718 | 10283.8 | 2591.98 | 3.61 | 394 | 324 | | | |

Table 1. Charges- commercial Fe-Ni recovery.

| Month | Ni(%) | Fetot(%) | SiO ₂ (%) | CaO(%) | MgO(%) | quantity(T |
|----------|-----------|-----------|----------------------|-----------|-------------|------------|
| january | 0.236 | 58.3 | 3.28 | 12.94 | 3.77 | N) 2520 |
| February | 0.22 | 56.4 | 3.4 | 13.4 | 3.6 | 2500 |
| March | 0.233 | 59.25 | 2.36 | 13.02 | 3.54 | 2500 |
| April | 0.24 | 58.74 | 1.63 | 12.74 | 3.58 | 2600 |
| May | 0.255 | 57.99 | 2.93 | 10.54 | 3.18 | 2269 |
| June | 0.312 | 60 | 2.78 | 9.99 | 3.06 | 3200 |
| TOTAL | 0.2519863 | 58.524877 | 2.7229309 | 12.038364 | 3.442544102 | 15589 |

Table 2. Converter slag analysis .

| Month | Ni(%) | Co(%) | S(%) | Si(%) |
|----------|-------|-------|------|-------|
| January | 25.44 | 1.03 | 0.12 | 0.01 |
| February | 24.25 | 0.91 | 0.09 | 0.01 |
| March | 25.21 | 0.96 | 0.11 | 0.01 |
| Aprile | 26.28 | 0.97 | 0.12 | 0.01 |
| May | 24.24 | 0.97 | 0.12 | 0.01 |
| June | 25.2 | 0.93 | 0.12 | 0.01 |

Table 3. Commercial Fe-Ni

By the above mentioned parameters we can see, one electric furnace is produced around 35000 t of converter slag per year, while with two electric furnaces, around 70000t converter slag per year. This quantity of slag, despite considerable amount of Ni 0,25 – 0,35% Ni and low amount of humidity 0,6% H₂O, because of technical reason wasn't used into the smelting process of obtaining Fe-Ni.

3. Processing of converter slag in smelting process of obtaining Fe-Ni

After additional crushing up to 20-50 mm, the converter slag has been processed in smelting process of obtaining Fe-Ni.

3.1 Treatment of converter slag in rotary kiln

As typical examples of industrial treatment of converter slag into the rotary kiln, we will present two cases as it follows: - Adding of converter slag without Albanian ore. - Adding of converter slag in combination with Albanian ore.

Working parameters of rotary kiln while operation with converter slag:

Dated on 12.07.2009, rotary kiln nr2 was operating as follow:

Capacity 70t/hr
Lignite 8,5%
Limestone 2,5%,

Ore composition:

Gllavica ore 47%
Cikatova ore 37,6%
Suka 1 ore 9,4%
Converter slag 6%
ε =100%

Charge composition:

Gllavica ore 32,9t
Cikatova ore 26,32t

Suka1ore 6,58t
 Converter slag 4,2t
 Lignite 5,9t
 Limstone 1,75t

Charge quantity 77,65t
 Chemical composition of ore, calcine, temperature of material and gases are given in table 4, 5 and 6.

| Ni % | Fe % | Co % | Cr ₂ O ₃ % | CaO % | MgO % | Al ₂ O ₃ % | SiO ₂ % | Fe ₂ O ₃ % | H.P % | Humidity % |
|------|-------|------|----------------------------------|-------|-------|----------------------------------|--------------------|----------------------------------|-------|------------|
| 0,92 | 24,21 | 0,06 | 1,94 | 7,30 | 6,69 | 2,79 | 30,34 | 34,62 | | 26,7 |
| 0,94 | 16,50 | 0,04 | 1,00 | 8,80 | 8,18 | 2,72 | 37,18 | 23,59 | | 29,30 |
| 1,22 | 18,08 | 0,05 | 1,06 | 2,26 | 8,26 | 3,32 | 46,00 | 25,85 | | 25,20 |

Table 4. Chemical composition of ore

| Ni % | Fe _{tot.} % | Co % | Cr ₂ O ₃ % | CaO% | MgO% | Al ₂ O ₃ % | SiO ₂ % | C _{fix} % | Fe _{met.} % | FeO % | Fe ₂ O ₃ % |
|------|----------------------|------|----------------------------------|------|-------|----------------------------------|--------------------|--------------------|----------------------|-------|----------------------------------|
| 1,17 | 17,87 | 0,04 | 1,37 | 4,25 | 12,29 | 2,46 | 44,74 | | | | |
| 1,06 | 16,08 | 0,03 | 1,10 | 6,15 | 11,14 | 1,30 | 42,27 | 3,50 | 0,19 | 13,14 | 8,15 |
| 1,18 | 16,62 | 0,04 | 1,16 | 3,33 | 12,50 | 1,93 | 44,16 | 2,40 | 0,19 | 12,42 | 9,72 |

Table 5. Chemical composition of calcine

OP – Optical Pyrometer.

Table 6. Temperature of material and gases inside rotary kiln

Working parameters of rotary kiln while the operation with converter slag and Albanian ore to:

Dated on 09.08.2009, rotary kiln nr2 was operating as follow:

Capacity 85t/hr
 Lignite 8,5%
 Limstone 3,5%

Ore composition:

Gllavica ore 52%
 Cikatova ore 32,135%
 Albanian ore 8,04%
 Filipinas ore 4,02%
 Converter slag 3,5% ε =100

Charge composition

Gllavica ore 44,46t
 Cikatova ore 27,314t
 Albanian ore 6,834t
 Filipinas ore 3,417t
 Converter slag 2,975t
 Lignite 7,225t
 Limstone 2,975t
 Charge quantity 95,2t

Chemical composition of ore, calcine, temperature of calcine and gases are given in table: 7, 8 and 9.

At the first example, for capacity of the rotary kiln of 70t/hr, in the charge was adding 6% cs/hr, or 4,2 tcs/hr $\times 0,995 \times 58,52 \%Fe_{total} = 2,44 Fe_{total}/hr$. The quantity of reducing CaCO₃ due to increasing of converter slag is 7.5% of the overall charge, or rotary kiln capacity of 70 t / hr: 70t/hr 7,5%=5,25 t CaCO₃ /hr, or considering the moisture of CaCO₃ from 0.85%, then 5,25 t CaCO₃ 0,9915 = 5,205t CaCO₃/hr. The practical results proved that≈60% of CaCO₃ dissociated in rotary kiln under reaction:

$CaCO_3 \rightarrow CaO + CO_2$
 $520 \text{ kgCaCO}_3/hr \times 60\% = 31223 \text{ kgCaCO}_3 \times 44/100 = 1374,12 \text{ kgCO}_2/hr,$
 Or $1374,12 \text{ kg CO}_2/hr \times 22,4/44 = 699,55 \text{ m}^3\text{NCO}_2/hr.$

Which means that for the capacity of 70t/hr of the rotary kiln has been produced less 699, 55 m³NCO₂/hr. For 1t dry ore has been produced less CO₂:
 $699,55 \text{ m}^3\text{NCO}_2/51,056 \text{ t DO} = 13,7 \text{ m}^3\text{NCO}_2/ \text{ t DO}.$

Quantity of Ni which is increasing inside of rotary kiln due to converter slag for 1t dry ore (DO) is:

$4,2tc. \text{ slag} \times 0,995/51,058 \text{ t DO} = 0,082tc. \text{ Slag/t DO}$
 $\times 0,253\%Ni = 0,000207,$
 or 0,207 kg Ni/t DO.

| Ni % | Fe % | Co % | Cr ₂ O ₃ % | CaO % | MgO% | Al ₂ O ₃ % | SiO ₂ % | Fe ₂ O ₃ % | H.P % | Humidity % |
|------|-------|------|----------------------------------|-------|------|----------------------------------|--------------------|----------------------------------|-------|------------|
| 1,11 | 15,79 | 0,04 | 0,95 | 5,77 | 8,87 | 6,75 | 42,51 | 22,57 | | 34,81 |
| 0,82 | 23,65 | 0,04 | 1,53 | 7,25 | 8,50 | 4,54 | 22,32 | 33,61 | | 22,85 |
| 1,33 | 17,70 | 0,06 | 1,15 | 5,42 | 8,07 | 3,01 | 40,77 | 25,31 | | 30,48 |

Table 7. Chemical composition of ore

| Ni % | Fetot. % | Co % | Cr2O3% | CaO % | MgO% | Al2O3% | SiO2% | Cfix% | Femet.% | FeO % | Fe2O3% |
|------|----------|------|--------|-------|-------|--------|-------|-------|---------|-------|--------|
| 1,08 | 20,33 | 0,05 | 1,44 | 4,04 | 11,12 | 2,82 | 44,66 | 1,75 | | | |
| 1,19 | 17,02 | 0,04 | 1,36 | 4,90 | 10,34 | 2,25 | 45,44 | 3,90 | 0,53 | 16,03 | 5,82 |
| 1,17 | 21,98 | 0,05 | 1,91 | 3,73 | 8,87 | 3,72 | 44,14 | 3,80 | 0,25 | 12,97 | 16,70 |

Table 8. Chemical composition of calcine.

| Temp. | M8 | M7 | M6 | M5 | M4 | M3 | M2 | M1 | G7 | G6 | G5 | G4 | G3 | G2 | G1 | G0 |
|-------|-----|-----|-----|-----|-----|-----|-----|-----|----|----|-----|-----|-----|-----|----|-----|
| OP °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C | °C |
| 760 | 826 | 516 | 822 | 594 | 342 | 331 | 244 | 160 | | | 839 | 589 | 509 | 448 | | 228 |
| 760 | 825 | 393 | 769 | 515 | 314 | 302 | 226 | 170 | | | 807 | 574 | 516 | 426 | | 234 |
| 756 | 816 | 302 | 660 | 401 | 293 | 310 | 266 | 205 | | | 740 | 568 | 517 | 415 | | 242 |

Table 9. Temperature of material and gases inside the rotary kiln

| Ni % | Fe % | Co % | Cr% | Si % | Cu % | S% | C% |
|-------|-------|------|------|------|------|------|------|
| 12,53 | 86,53 | 0,42 | 0,13 | 0,04 | 0,02 | 0,57 | 0,19 |
| 12,36 | 86,79 | 0,42 | 0,12 | 0,01 | 0,03 | 0,57 | 0,19 |
| 12,43 | 87,11 | 0,42 | 0,12 | 0,02 | 0,03 | 0,57 | 0,20 |

Table 10. Chemical composition of crude Fe-Ni

| Ni % | SiO ₂ % | Fe % | FeO% | CaO % | MgO % | MnO % | Cr2O3 % | Al ₂ O ₃ % |
|------|--------------------|-------|-------|-------|-------|-------|---------|----------------------------------|
| 0,07 | 56,10 | 14,83 | 19,13 | 6,82 | 16,39 | 0,17 | 2,20 | 2,23 |
| 0,07 | 56,41 | 14,44 | 18,62 | 6,75 | 16,97 | 0,18 | 2,14 | 2,18 |
| 0,07 | 54,71 | 14,07 | 18,15 | 6,51 | 16,38 | 0,19 | 1,92 | 2,14 |

Table 11. Chemical composition of electric furnace slag

| Ni % | Fe % | Co % | Cr% | Si % | Cu % | S% | C% |
|-------|-------|------|------|------|------|------|------|
| 12,53 | 86,53 | 0,42 | 0,13 | 0,04 | 0,02 | 0,57 | 0,19 |
| 12,98 | 84,46 | 0,48 | 0,32 | 1,18 | 0,03 | 0,57 | 0,41 |
| 13,32 | 84,08 | 0,48 | 0,30 | 1,83 | 0,03 | 0,60 | 0,38 |

Table 12. Chemical composition of crude Fe-Ni

| Ni % | SiO ₂ % | Fe % | FeO% | CaO % | MgO % | MnO % | Cr ₂ O ₃ % | Al ₂ O ₃ % |
|------|--------------------|-------|-------|-------|-------|-------|----------------------------------|----------------------------------|
| 0,06 | 60,89 | 15,17 | 19,56 | 6,48 | 11,68 | 0,35 | 1,54 | 2,24 |
| 0,08 | 60,69 | 15,74 | 20,30 | 6,37 | 11,40 | 0,37 | 1,51 | 2,23 |
| 0,07 | 59,15 | 15,15 | 20,31 | 6,26 | 11,25 | 0,37 | 1,48 | 2,20 |

Table 13. Chemical composition of electric furnace slag

| Ni % | SiO ₂ % | Fe % | FeO% | CaO % | MgO % | MnO % | Cr ₂ O ₃ % | Al ₂ O ₃ % |
|------|--------------------|-------|-------|-------|-------|-------|----------------------------------|----------------------------------|
| 0,07 | 56,10 | 14,83 | 19,13 | 6,82 | 16,39 | 0,17 | 2,20 | 2,23 |

Table 14. Chemical composition of electric furnace slag with increasing quantity by 5,43% F_{total}

| Ni % | SiO ₂ % | Fe % | FeO% | CaO % | MgO % |
|------|--------------------|------|-------|-------|-------|
| 0,07 | 58,49 | 9,41 | 12,08 | 7,11 | 17 |

Table 15. Chemical composition of electric furnace slag with less of 5,43 % F_{total}

| Ni % | SiO ₂ % | Fe % | FeO% | CaO % | MgO % | MnO % | Cr ₂ O ₃ % | Al ₂ O ₃ % |
|------|--------------------|-------|-------|-------|-------|-------|----------------------------------|----------------------------------|
| 0,06 | 60,89 | 15,17 | 19,56 | 6,48 | 11,68 | 0,35 | 1,54 | 2,24 |

Table 16. Chemical composition of electric furnace slag with increasing quantity by 3,27% F_{total}

| Ni % | SiO ₂ % | Fe % | FeO% | CaO % | MgO % |
|------|--------------------|------|-------|-------|-------|
| 0,07 | 61,2 | 11,4 | 14,65 | 6,61 | 11,93 |

Table 17. chemical composition of electric furnace slag with less of 3,27% F_{total}

3.2 Treatment of converter slag in electric furnace

Products of electric furnace for the first example are given in table 10, 11, while for second example in table 12, 13.

Additional quantity of converter slag, is increasing the quantity of F_{total} into electric furnace slag and Ni quantity in crud Fe-Ni as follow:

Chemical composition of electric furnace slag with increasing of F_{total} by 5,43% is given in table 14, while with less F_{total} by 5,43% is given in table 15.

Chemical composition of electric furnace slag with increasing of F_{total} by 3,27% is given in table 16, while with less F_{total} by 3,27% is given in table 17.

For first example:

Increasing quantity of F_{total} inside electric furnace slag is:

$51,058t \text{ DO/hr} \times 88\% = 44,93t_{\text{elec.fur.slrag}}/\text{hr}$,
and $2,44t \text{ } F_{total} /\text{hr} \times 100/44,93t_{\text{elec.fur.slrag}} /\text{hr} = 5,43\%F_{total}$.

The practical results proved that $\approx 40\%$ of CaCO_3 has dissociated inside electric furnace under reaction.



$520 \text{ kg } \text{CaCO}_3 /\text{hr} \times 40\% = 2082 \text{ kg } \text{CaCO}_3/\text{hr} \times 44/100 = 916,08 \text{ kg } \text{CO}_2/\text{hr}$,
or $916,08 \text{ kgCO}_2/\text{hr} \times 22,4/44 = 466,368 \text{ m}^3\text{NCO}_2/\text{hr}$

Which means that for the capacity of 70t/hr of the rotary kiln inside the electrical furnace has been produced less $466,368 \text{ m}^3\text{NCO}_2/\text{hr}$.

For 1t dry ore has been produced less CO_2 :

$$466,368 \text{ m}^3\text{NCO}_2/51,056t \text{ DO} = 9,13 \text{ m}^3\text{NCO}_2/t \text{ DO}.$$

4. RESULTS AND DISCUSSION

Results from first example: .

Increased quantity of $F_{total} /t \text{ DO}$: $0,0479t \text{ } F_{total} /t \text{ DO}$.

Decreasing quantity of CO_2 in rotary kiln gases: $13,7 \text{ m}^3\text{NCO}_2/t \text{ DO}$.

Decreasing quantity of CO_2 in electric furnaces gases: $9,13 \text{ m}^3\text{NCO}_2/t \text{ DO}$.

Increased quantity of Ni from converter slag: $0,207 \text{ kg Ni/t DO}$.

Increased quantity of F_{total} in electric furnace slag: $5,43\% F_{total}$.

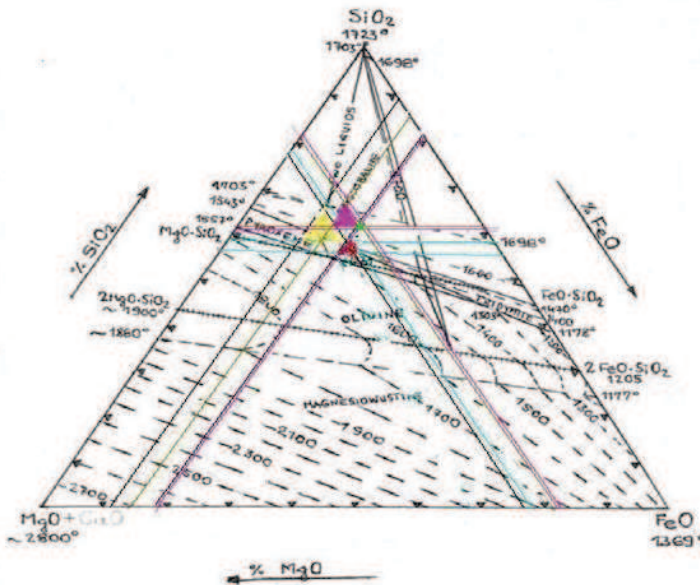


Fig 1. Three components diagram: $MgO+FeO+SiO_2$

Results from second example:

Increased quantity of Fe_{total} /t DO: 0,028t Fe_{total} /t DO.

Decreasing quantity of CO_2 in rotary kiln gases: 12,25 m^3NCO_2 /t DO

Decreasing quantity of CO_2 in electric furnaces gases: 8,16 m^3NCO_2 /t DO.

Increased quantity of Ni from converter slag: 0,123 kg/t DO

Increased quantity of Fe_{total} in electric furnace slag: 3,27% Fe_{total}

From three components diagram (fig.1) we can compare the liquid temperature of electric furnace slag for 2 examples, depending from chemical composition of slag

First example – Operation only with converter slag

a). With converter slag

b). Without converter slag:

$SiO_2 = 56,1\%$; $SiO_2 = 58,49\%$; $MgO + CaO = 16,39\% + 6,82\% = 23,21\%$; $MgO + CaO = 17\% + 7,11\% = 24,11\%$; $FeO = 19,13\%$; $FeO = 12,08\%$; $TI \approx 1500^\circ C$. $TI \approx 1600^\circ C$.

Second example – converter slag and Albanian ore

a). With converter slag and Albanian ore: b). Without converter slag:

$SiO_2 = 60,89\%$; $SiO_2 = 61,2\%$; $MgO + CaO = 11,68\% + 6,48\% = 18,16\%$; $MgO + CaO = 11,93\% + 6,61\% = 18,54\%$; $FeO = 19,56\%$; $FeO = 14,65\%$; $TI \approx 1650^\circ C$ $TI \approx 1680^\circ C$.

By facilitation of the process for diffusion of hot gases, through the material inside the rotary kiln has improved roasting process of calcine and has reduced quantity of sticking calcine for 50%. Such improvement of technological process of the roasting, has created condition for increasing of quantity and quality of calcine, producing more calcine, with temperature over $750^\circ C$ (that in the past was around $450^\circ C$) and pre-reduction degree about 56% (that was almost 0%) that has indicated directly in decreasing of black oil consumption up to 53kg/t calcine and decreasing of electrical specific consumption in electrical furnace from 604KWh/t calcine that was in 2007 in 514KWh/t calcine in July 2009.

Is decreased additional quantity of sterile material $CaCO_3$ in charge from 10% up to 3,5%- 2,5%, that has indicated positively in increasing of percentage of Ni in calcine for 0,207kgNi/t DO (in the first example), for 0,123kgNi/t DO (in the second example) and in decreasing of quantity of CO_2 in rotary kiln gases for 13,7 m^3NCO_2 /t DO (in the first example), for 12,25 m^3NCO_2 /t DO (in second example) and in the electric furnace gases for 9,13 m^3NCO_2 /t DO (in the

first example) and $9,13\text{m}^3\text{NCO}_2/\text{t DO}$ (in second example).

By increasing quantity of Fe_{total} in electric furnace slag for 5, 43% (for first example) and 3,72% (for second example) in both cases have been created satisfactory condition, that by decreasing of acidity of electric furnace slag, to decrease the smelting point of it and such case to have a technological process so equilibrium in electric furnace, with continuous flow of slag (approximately is preserved condition: $\text{FeO} + \text{MgO} + \text{CaO} \approx 40\%$). In the first example:

$$\text{FeO} + \text{MgO} + \text{CaO} = 42,34\%$$

and in the second example:

$$\text{FeO} + \text{MgO} + \text{CaO} = 37,72\%$$

If we would operate without converter slag for two ceases, we would have these parameters:

$$\text{FeO} + \text{MgO} + \text{CaO} = 36,19\% \text{ (for first example)}$$

and

$$\text{FeO} + \text{MgO} + \text{CaO} = 33,19\% \text{ (for second example)}.$$

Has been increased considerably the recovery of Ni, from 57, 5% in the 2007, 91, 19% in 2008 and in July 2009 87, 3%.

5. CONCLUSIONS

Industrial investigation has been done at smelting complex of Newco Ferronikeli Complex in Drenas.

By industrial experiments has been proved, that converter slag could be treated without difficulty in metallurgical process of obtaining Fe-Ni, recycling it into the smelting process as single component or in mixture with Albanian ore.

Recycling of converter slag in smelting process of obtaining Fe-Ni, except improvement of metallurgical process has been indicated positively in decreasing of pollution for environment as well. In the future should investigate new methods for further improving of metallurgical process of obtaining Fe-Ni, eliminating at all using of CaCO_3

6. LITERATURE

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